

ARIZONA STATE UNIVERSITY

#### LightWorks®

#### **Unlocking Solar Thermochemical Potential Workshop**

#### **Panelist Ellen B. Stechel**

#### ASU LightWorks<sup>®</sup>, Arizona State University, Tempe AZ, USA

#### **COLLABORATORS**

Arizona State University: Shuguang Deng, Ivan Ermanoski, James E. Miller, Ryan Milcarek, Chris Muhich Georgia Institute of Technology: Peter Loutzenhiser

National Renewable Energy Laboratory: David Ginley, Karen Heinselman, Bob Bell, Phil Parilla, Dan Plattenberger, Sarah Shulda Oregon State University: Brian Fronk

Princeton University: Emily Carter (now UCLA), Sai Gautam (now IISc), Rob Wexler

Sandia National Laboratories: Kevin Albrecht, Andrea Ambrosini, Evan Bush, Eric Coker, Matt Kury, Tony McDaniel

**Siemens Corporation**: Ioannis Akrotirianakis, Arindam Dasgupta, Ayse Parlak

Southwest Research Institute: Stefan Cich, Josh Neveu

**SETO NOVEMBER 2020** 

UNLOCKING THERMOCHEMICAL POTENTIAL

ASU LIGHTWORKS®

ELLEN B STECHEL



STA

# **SOLAR THERMOCHEMISTRY CONTEXT**

- Chemistry: Making and breaking bonds
- Thermochemistry combines the concepts of thermodynamics with the idea of energy in the form of chemical bonds.
- Thermochemistry is a branch of thermodynamics that is the study of heat generated (exotherm) or consumed (endotherm) in a chemical reaction.
- Solar: Source of heat particularly to promote the endothermic reactions
   Heat can also accelerate reactions even if exothermic

   High temperature can be sustainable when enabled by solar fuels or concentrated solar
- Deep Decarbonization: Applications to a large range of carbon-intensive sectors
- Dispatchability: Enable deeper penetration of renewables with less curtailment





## SOME (MANY) APPLICATIONS OF SOLAR THERMOCHEMISTRY



3



ARIZON

Α

S T A



### SOME THERMOCHEMICAL REACTIONS OF INTEREST

**Thermochemical Energy Storage** 

**Redox: Oxygen Shuttle** 

$$\frac{1}{\delta} MO_x \rightarrow \frac{1}{\delta} MO_{x-\delta} + \frac{1}{2}O_2$$
  

$$H_4 + \frac{1}{\delta} MO_x \rightarrow \frac{1}{\delta} MO_{x-\delta} + CO + 2H_2$$
  

$$H_2 + \frac{1}{\delta} MO_x \rightarrow \frac{1}{\delta} MO_{x-\delta} + H_2O$$

$$\frac{1}{\delta}MO_{x-\delta} + H_2O \rightarrow \frac{1}{\delta}MO_x + H_2$$
$$\frac{1}{\delta}MO_{x-\delta} + CO_2 \rightarrow \frac{1}{\delta}MO_x + CO$$
$$\frac{1}{\delta}MO_{x-\delta} + Air \rightarrow \frac{1}{\delta}MO_x + N_2$$

 $\frac{1}{\delta}MO_{x} \stackrel{\rightarrow}{\leftarrow} \frac{1}{\delta}MO_{x-\delta} + \frac{1}{2}O_{2}$   $\overrightarrow{CaCO_{3}} \stackrel{\rightarrow}{\leftarrow} CaO + CO_{2}$   $\overrightarrow{Ca(OH)_{2}} \stackrel{\rightarrow}{\leftarrow} CaO + H_{2}O$ 

Thermochemical Ammonia  $\frac{1}{\gamma}MN_{x} + \frac{3}{2}H_{2} \rightarrow \frac{1}{\gamma}MN_{x-\gamma} + NH_{3}$   $\frac{1}{\gamma}MN_{x-\gamma} + \frac{1}{2}N_{2} \rightarrow \frac{1}{\gamma}MN_{x}$  Gasification

 $C_{x}H_{2y}O_{z} + (x-z)H_{2}O \rightarrow (y+x-z)H_{2} + xCO$ 

#### Reforming

 $CH_4 + H_2O \rightarrow CO + 3H_2$  $CH_4 + CO_2 \rightarrow 2CO + 2H_2$ 

Each application presents different challenges at the system level but many commonalities too Important to co-optimize materials, reactors, desired operating range, and the systems

ASU LIGHTWORKS<sup>®</sup>



### **KEY RISKS OFTEN OVERLOOKED EARLY IN THE DEVELOPMENT PROCESS**

Having a good technical risk assessment and review formalism

Identifying all the possible failure modes – what would keep the material, the functional components, interfaces, or the system from working as intended?

Potential Failure mode	Potential Effect(s) of Failure	Severity (1-10)	Potential Mechanism(Causes) of Failure Mode	Occurrence (1-10)	Current Design Features/ Controls	Detection (1-10)	Risk Priority Number (RPN)	Recommended Actions
---------------------------	-----------------------------------	--------------------	---	----------------------	---	---------------------	-------------------------------------	------------------------

- The FMEA (Failure Mode Effects & Analysis) is a collaborative exercise and works best with a diverse team
- Can be effective at identifying and mitigating or eliminating risks •
- Applies broadly, e.g., to design of functional materials, components, interfaces, and the full system .
- Early on sufficient to drive the RPN = Severity  $\times$  Occurrence  $\times$  Detectability < 100

**SETO NOVEMBER 2020** 

UNLOCKING THERMOCHEMICAL POTENTIAL

**ASU LIGHTWORKS®** 



### FISHBONE DIAGRAM CAN AID IN ASSESSING RISKS





### **VERY IMPORTANT RISK IS GETTING THE SCALE RIGHT**

Solar vs. Nuclear capacity built per decade



- Flexibility, Adaptability
- Faster Learning
- Less Investor Risk
- Start generating revenue more quickly
- Matching scale with downstream processing, e.g., syngas → jet fuel

7



### RISK OF NOT OPTIMIZING FOR THE THERMODYNAMICS BECAUSE OF HIGH TEMPERATURE

Challenging but not necessarily a show-stopper: Know the difference between an engineering challenge that might have analogs in other applications and show-stoppers

#### **High Temperature Industrial Processes**

#### Fusing quartz under H<sub>2</sub>/O<sub>2</sub> flame: T~1700°C









### REGIONS WITH AN EXCELLENT SOLAR RESOURCE STILL MUST COPE WITH SUBSTANTIAL VARIABILITY



- Must consider the impact of the variability
- Determine performance off design point
- Might have both supply and demand variability (as with electricity).
- There may be consequences for downstream (off-sun) processes to consider

9



### COST DRIVERS ARE TECHNICAL METRICS IMPORTANT AT ALL PHASES OF DEVELOPMENT

- Total Project Investment or CAPEX: Cx ( $\frac{k}{W}$ )
- Capacity Factor: *CF* (Between 0 and 1)
- Energy Utilization:  $eU(kWh/kg) \propto 1/Efficiency$
- Cost of energy plus variable O&M: Ce \$/kWh
- Annualized cost factor (financial plus fixed O&M):  $crf(\frac{1}{vr})$

$$\frac{\$}{kG_{Product}} = eU \times \left\{ \frac{Cx \times crf}{CF \times 8760 \ hr/yr} + Ce \right\}$$

Power Density: kW/L Measure of compactness

Balancing solar constraints and balance of system – won't necessarily eliminate cost of energy

May not match up on CF either



#### STA ARIZON Α

### **Acknowledgments**



Work is funded in part by the U.S. DOE SETO Award Number 08991 Weekly and Seasonal Storage

- ASU, Siemens Corporation, U. of Oregon, Sandia 0 National Labs, and Southwest Research Institute
- DOE Project Manager: Matthew Bauer 0

Award Number 34250 Solar Thermochemical Ammonia Production

- Sandia National Labs (SNL), ASU, and Georgia 0 Institute of Technology
- DOE Project Manager: Levi Irwin



Work is funded in part by the U.S. DOE FCTO Award Number 08090 Materials Discovery Solar thermochemical water splitting

- ASU and Princeton (collaborative support 0 from SNL and NREL)
- DOE Project Manager: Katie Randolph 0

Award Number 08090 Materials Discovery – Solar thermochemical water splitting

**ASU LIGHTWORKS®** 

- Nel Hydrogen, ASU, Caltech, and PNNL
- DOE Project Manager: Katie Randolph

11

ELLEN B STECHEL

UNLOCKING THERMOCHEMICAL POTENTIAL



INLET TEMPERATURE ON JET ENGINES CONTINUOUSLY INCREASING



The evolution of allowable gas temperature at the entry to the gas turbine and the contribution of superalloy development, film cooling technology, thermal barrier coatings and (in the future) ceramic matrix composite (CMC) air foils and perhaps novel cooling concepts.

Ζ

R