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(54) **SOLAR SYSTEM FOR REPRODUCING THE EFFECT OF A COMBUSTION FLAME**

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F24J 2/24 (2006.01)
F24J 2/10 (2006.01)

(52) **U.S. Cl.**

CPC . **F24J 2/07** (2013.01); **F24J 2/24** (2013.01);
F24J 2/10 (2013.01); **Y02E 10/41** (2013.01);
Y02E 10/44 (2013.01); **Y02P 80/24** (2015.11)

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See application file for complete search history.

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(57) **ABSTRACT**

The present invention relates to a solar system for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterized in that it comprises:

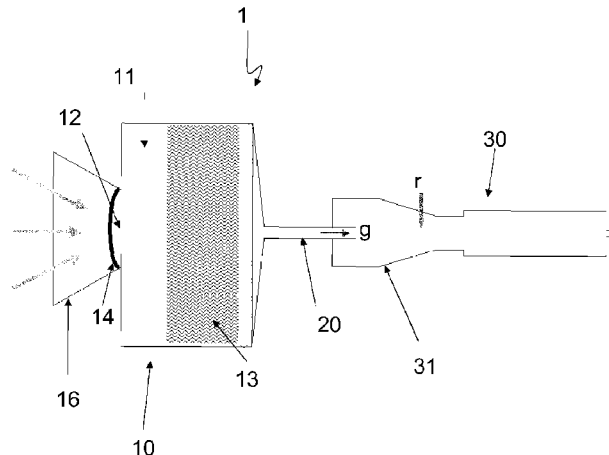
a solar receiver exposed to concentrated solar radiation, in which heat transfer fluid (liquid or gas) is brought to high temperature;

at least one high-temperature chamber in which said high-temperature industrial process is performed;

injection means of the heat transfer fluid in the form of a gas jet reproducing a combustion flame in the at least one high-temperature chamber.

The present invention also relates to a process for providing volumetric energy reproducing the effect of a combustion flame for this purpose.

20 Claims, 7 Drawing Sheets



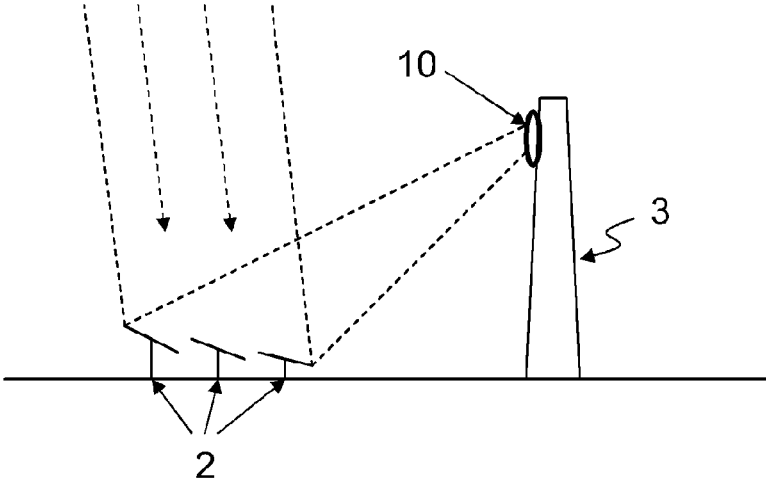


FIG. 1

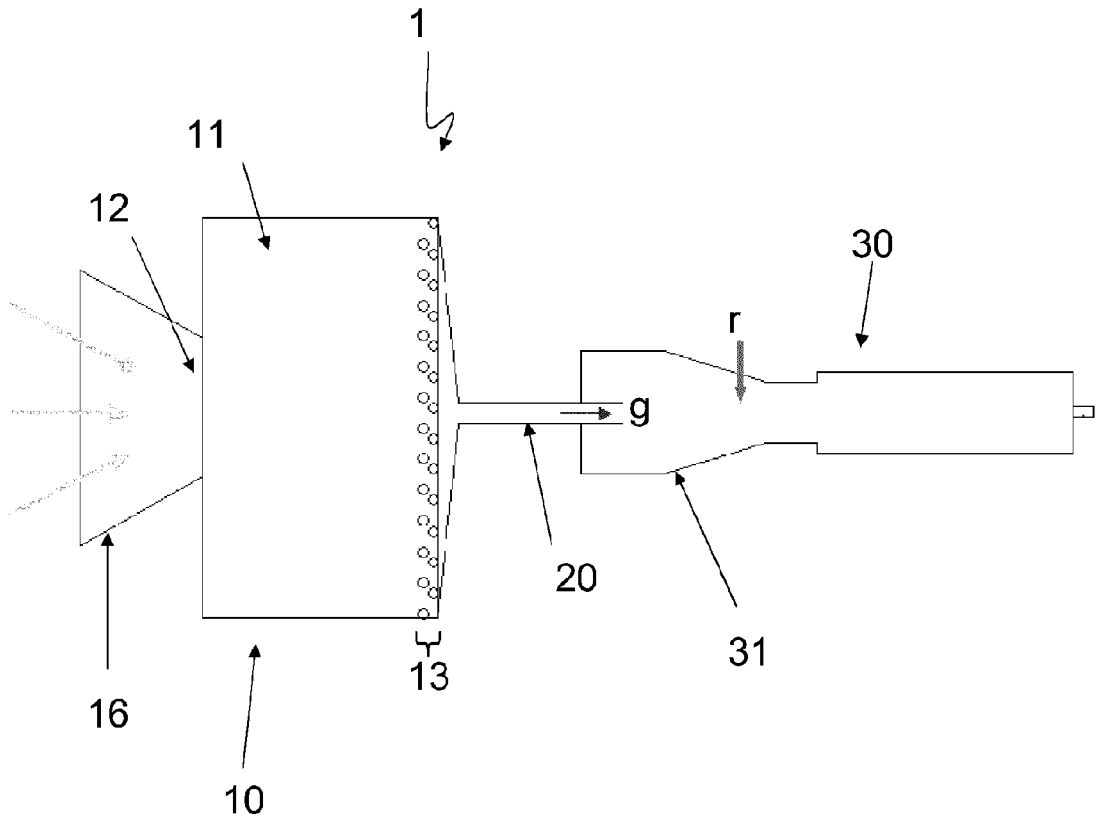


FIG. 2

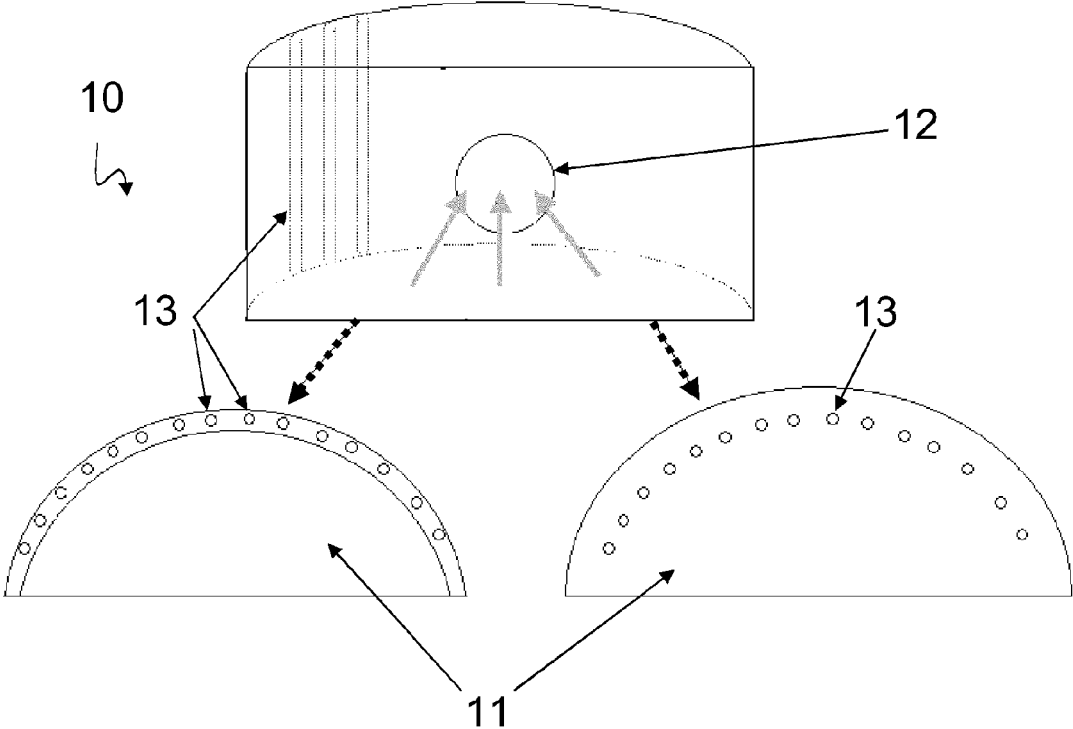


FIG. 3a-c

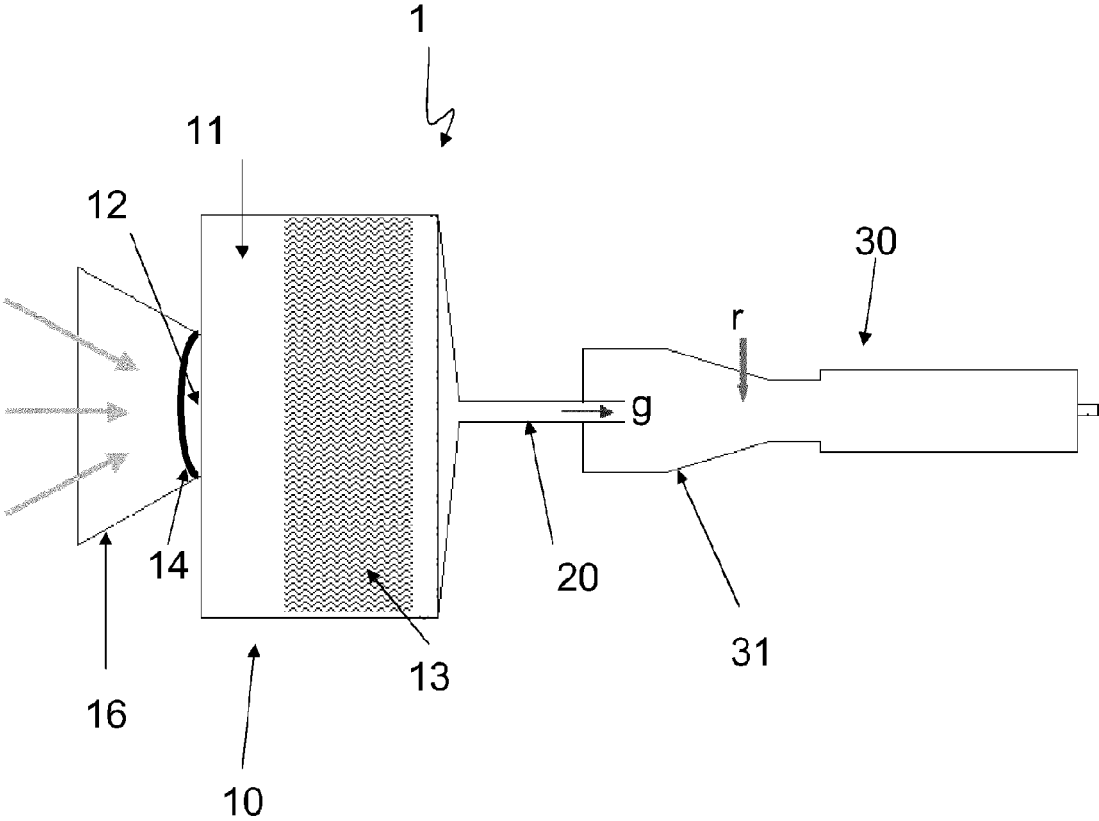


FIG. 4

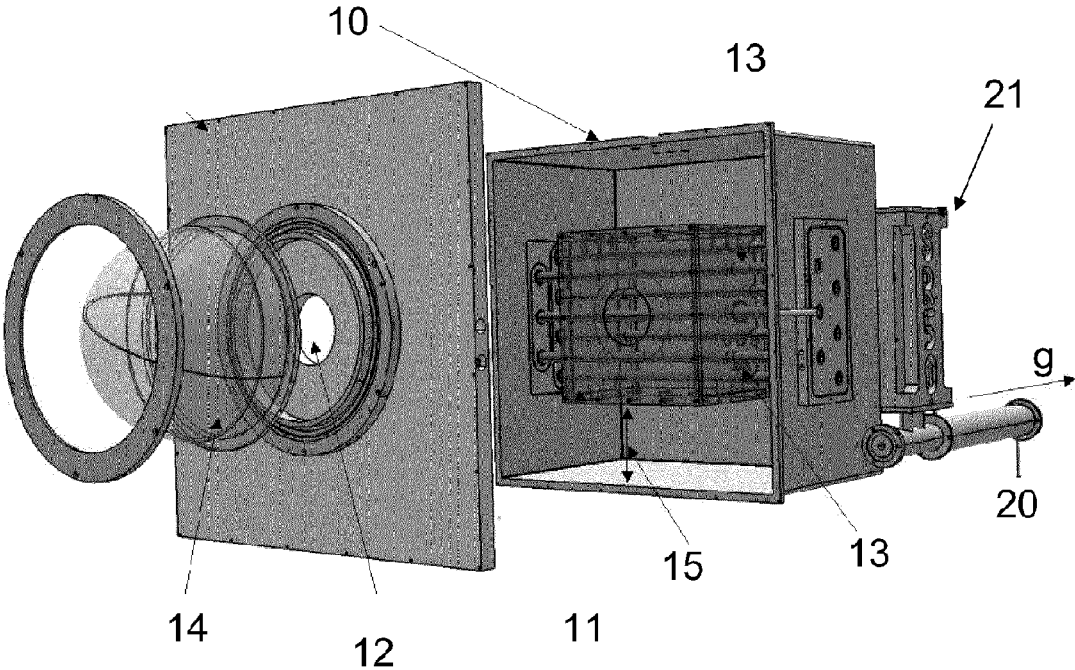


FIG. 5

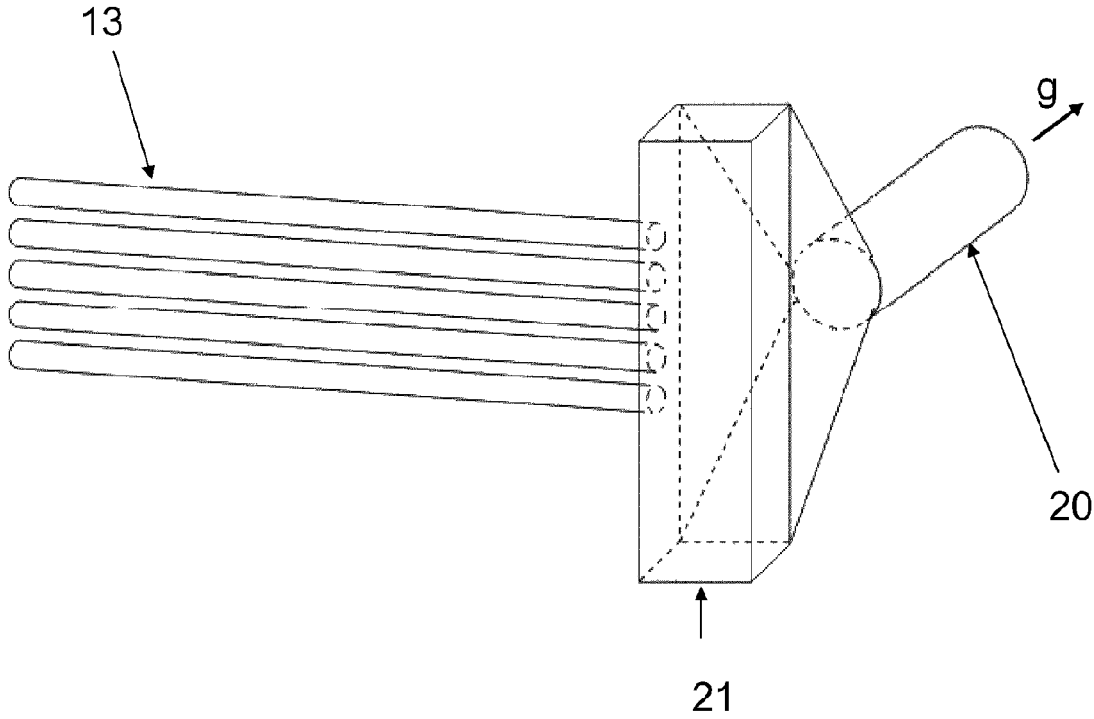


FIG. 6

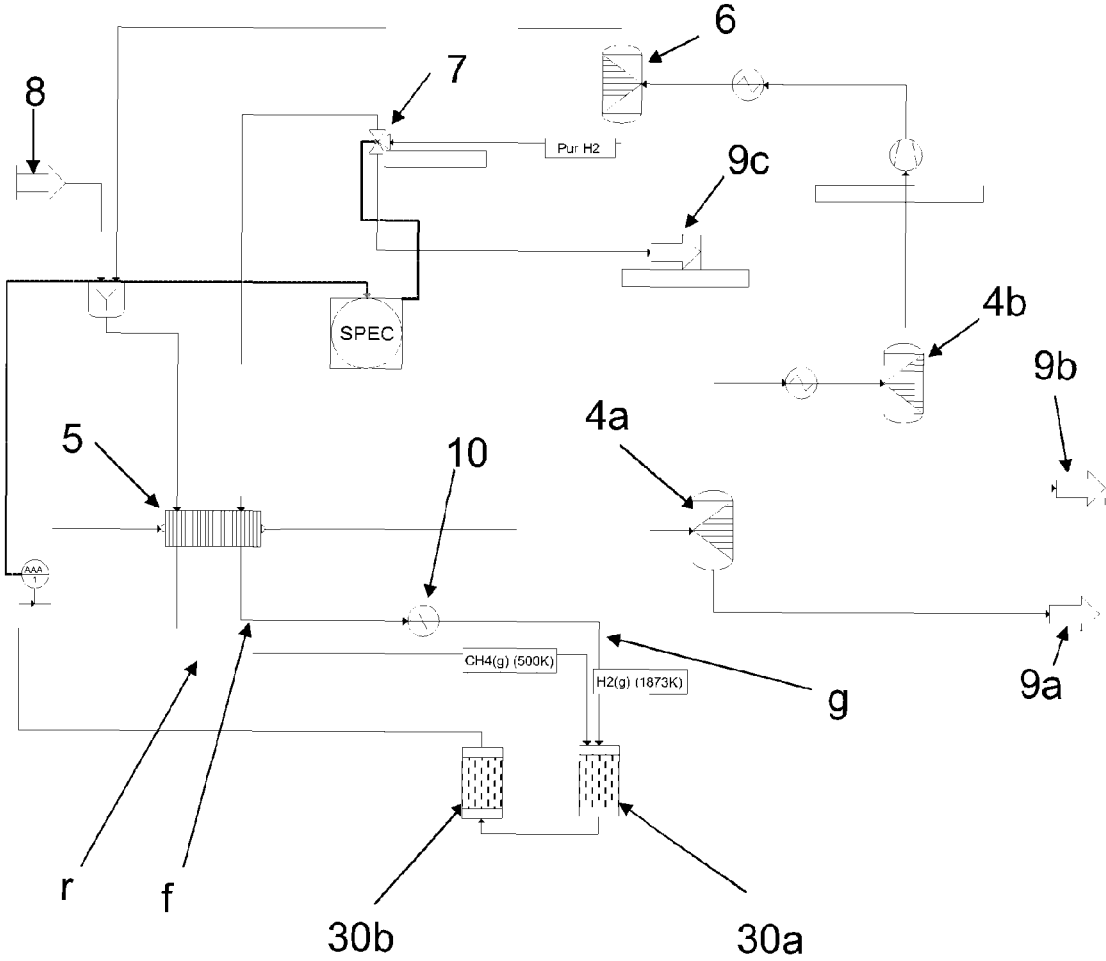


FIG. 7

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SOLAR SYSTEM FOR REPRODUCING THE EFFECT OF A COMBUSTION FLAME

CROSS REFERENCE TO RELATED APPLICATIONS

The present application is a national phase entry under 35 U.S.C. §371 of International Application No. PCT/EP2012/056129, filed Apr. 4, 2012, published in French, which claims priority from French Patent Application No. 1152862, filed Apr. 4, 2011, the disclosures of which are incorporated by reference herein.

GENERAL TECHNICAL FIELD

The present invention relates to the field of high-temperature solar thermal and thermochemical systems.

More precisely, it relates to a solar system for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process by means of transfer fluid.

PRIOR ART

Numerous industrial processes, for example the furnace process for producing carbon black, need a large input of heat energy to a volume of material to be treated. Required temperatures reach 2000° C.

Currently, these high temperatures are produced almost exclusively by the combustion of fossil resources, hydrocarbons in particular. In fact with the plasma process, <<flame combustion>> is one of the only ways to obtain volumetric thermal input (i.e. heat brought to a volume) at the industrial level. The name <<flame temperatures>> also designates those temperatures to which these processes are subjected.

These classic techniques have provided proof but best need a large electricity supply, if not consumption of fossil resources, and cause greenhouse gas emissions and/or pollutants (NOx, SOx, particles).

It would be preferable to have an alternative to combustion flame to be obtained from renewable energy only, and solar energy in particular.

In fact, so-called concentration solar systems enable conversion of solar radiation to heat energy, which is generally used for electricity production.

One of these concentration systems is the tower power plant. Such a system comprises a tower and a field of mobile mirrors called heliostats, which concentrates the radiation onto a reduced area at the apex of the tower.

This zone lit by the concentrated radiation receives several hundred times the direct solar irradiation and is equipped with a device called a solar receiver the function of which is to transmit this energy to a fluid (liquid or gas) which circulates there, generally water steam turbined to produce electricity. FIG. 1 illustrates a thermodynamic tower power plant comprising a tower 3, a field of heliostats 2, and the solar receiver 10.

The solar receiver described in patent application US 2010/0237291 comprises a cavity traversed by a plurality of tubes in which chemical compounds circulate which can react in so-called endothermic reactions, that is, needing high temperatures usually employed under the effect of a combustion flame (cracking of methane, for example). The external wall of the tubes receives the concentrated solar radiation, which brings it to high temperature: the tubes act as transfer wall. However, the thermal input is not volumet-

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ric input but surface input: the heat is transferred to the chemical reagents by contact with the internal wall of the tubes. This surface thermal input reproduces the effect of a combustion flame poorly, since chemical reaction takes place mainly at the level of the walls. This diminishes the yield and causes the appearance then growth of deposits of solid residue on the walls (in this case pure carbon in the case of a cracking process), which rapidly makes the receiver unusable: deposits decrease the thermal input and obstruct the tubes.

Alternatively, international patent application WO 03/049853 proposes a solar receiver (a porthole allows return of the concentrated solar radiation), in which the chemical compounds to be heated circulate directly. A cloud of solid absorbent microparticles (carbon black smoke, for example) is suspended in the receiver. It is these particles which are irradiated by the concentrated solar radiation and diffuse heat energy which they receive to carry out the endothermic reaction. Heat input this time is definitely volumetric (thermal exchange takes place rigorously on the surface of particles, but by relatively even diffusion of the particles in the volume it is assimilable to volumetric thermal input), and best reproduces the effect of a combustion flame. However, performances of this type of solar receiver are very limited: in fact, particles rapidly tend to deposit on the porthole and cloud it. Recovering reaction products is also complex as the resulting gas has to be filtered for isolating and recovering microparticles without contaminating the products. Industrial interest is therefore low.

Current technologies are yet to offer a viable industrial solar alternative to the combustion flame for high-temperature industrial processes.

PRESENTATION OF THE INVENTION

According to a first aspect, the present invention therefore relates to a solar system for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises:

a solar receiver exposed to concentrated solar radiation, in which heat transfer fluid (liquid or gas) is brought to a high temperature;

at least one high-temperature chamber in which said high-temperature industrial process is performed;

injection means of the heat transfer fluid in the form of a gas jet reproducing a combustion flame in the at least one high-temperature chamber.

According to other advantageous and non-limiting characteristics:

the solar receiver comprises a cavity, the cavity being provided with an opening transparent to concentrated solar radiation, and at least one solar absorption element irradiated by the concentrated solar radiation through the opening;

the solar absorption element or the solar absorption elements are tubes and/or ducts lining at least one wall of the cavity;

the heat transfer fluid circulates in the solar absorption element or the solar absorption elements, the heat transfer fluid being brought to a high temperature by heat transfer in contact with the internal surface of the solar absorption elements;

the heat transfer fluid circulates in the cavity, the opening being covered with a porthole transparent to concentrated and sealed solar radiation, and the heat transfer fluid being

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brought to high temperature by heat transfer in contact with the external surface of the solar absorption element or the solar absorption elements;

the cavity and the solar absorption element or the solar absorption elements are made of ceramic or graphite;

the solar receiver is traversed by a duct in which the heat transfer fluid is in motion, the heat transfer fluid being brought to a high temperature by heat transfer in contact with the internal surface of a wall of the duct whereof the external surface is irradiated by the concentrated solar radiation;

the heat transfer fluid is injected under pressure into the solar receiver;

the injection means of the heat transfer fluid in the high-temperature chamber consist of a conduit via which the high-temperature heat transfer fluid escapes from the solar receiver to the high-temperature chamber under the effect of the internal pressure in the solar receiver;

the pressure in the high-temperature chamber is less than the pressure in the solar receiver, the form of gas jet being caused by detente of the heat transfer fluid as it leaves the injection means:

the temperature of the high-temperature heat transfer fluid is between 1000° C. and 2500° C.

the high-temperature chamber is a furnace, and the high-temperature industrial process is a process for obtaining metallic or ceramic material;

the high-temperature chamber is a chemical reactor, and the high-temperature industrial process is an endothermal chemical reaction;

the heat transfer fluid comprises chemically inert gas and/or a reagent of said endothermal chemical reaction and/or a product of said endothermal chemical reaction;

at least one reagent of said endothermal chemical reaction is injected into the high-temperature reactor at the level of an injection zone of the high-temperature heat transfer fluid;

the endothermal chemical reaction is the cracking of methane;

the heat transfer fluid is dihydrogen, methane being injected into the high-temperature reactor;

the system comprises a plurality of high-temperature chambers, each being a chemical reactor, the products of the nth reactor being injected into the n+1st reactor.

A second aspect of the invention relates to a process for volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises steps of:

irradiation of a solar receiver in which heat transfer fluid circulates by concentrated solar radiation to bring the heat transfer fluid to high-temperature;

injection of the heat transfer fluid into a high-temperature chamber in the form of a gas jet reproducing a combustion flame;

performing said high-temperature industrial process in the high-temperature chamber under the effect of the combustion flame produced.

PRESENTATION OF FIGURES

Other characteristics and advantages of the present invention will emerge from the following description of a preferred embodiment. This description will be given in reference to the attached drawings, in which:

FIG. 1 previously described is a drawing of a known solar thermodynamic tower power plant;

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FIG. 2 is a drawing of an embodiment of the solar system according to the invention;

FIG. 3a is a view in perspective of a solar receiver used by the system according to the invention, and FIGS. 3b and 3c are two views in section of two embodiments of this solar receiver;

FIG. 4 is a drawing of another embodiment of the solar system according to the invention;

FIG. 5 is a drawing of a particularly advantageous embodiment of a solar receiver of the solar system according to the invention;

FIG. 6 is a drawing of an embodiment of a collector of the solar system according to the invention;

FIG. 7 is a fluid circulation plan (Prosim® software) in a particularly advantageous embodiment of the system according to the invention.

DETAILED DESCRIPTION

General Architecture and Principle

In reference to the drawings and first to FIG. 2, the solar system 1 according to the invention comprises three main parts: a solar receiver 10, a high-temperature chamber 30, and fluid injection means 20 from the solar receiver 10 to the high-temperature chamber 30.

The <<high temperatures>> to be mentioned in the present description correspond to temperatures which can almost be attained only by a combustion flame or plasma, and not by simple electrical resistances. These high temperatures are typically above 1000° C. and if needed can reach 2000° C. or even 2500° C.

As is seen from FIG. 2, the solar receiver is exposed to concentrated solar radiation. As explained previously, concentration of the solar radiation means focussing solar radiation received by a vast surface onto a small surface using optical systems: fields of mirrors (heliostats), large parabolic mirror, lenses, etc. The energy received by the receiver 10 equals total solar energy incident to optical losses, for example close to energy captured by the total surface of the field of heliostats 2 in FIG. 1.

Heat transfer fluid f circulates in the receiver 10, this fluid being brought to high temperature under the effect of the concentrated solar radiation. Different types of fluids and different receiver geometries will be described in detail below.

The high-temperature chamber 30 is as such the site of a high-temperature industrial process. As will be explained hereinbelow, numerous industrial processes can be employed within the scope of the present invention, in particular any process needing volumetric energy, and especially if this input is generally made by a combustion flame.

The core of the invention is injection via the injection means 20 of the heat transfer fluid f in the form of a gas jet g into the at least one high-temperature chamber 30. In fact, a flame comprises only final high-temperature gas products or intermediate products of the combustion reaction which engendered it (generally CO₂, H₂O), the luminous aspect of the flame due to excitation of electrons of these gases. A high-temperature gas jet can therefore be equivalent to a combustion flame. Therefore, injecting the heat transfer fluid f heated at the level of a fine opening of the chamber 30 at a sufficient rate simulates a burner and a combustion flame is reproduced.

Architecture of Solar Receiver

The expert knows numerous types of solar receivers 10. It is On noted that the invention is not limited to any type of receiver in particular, but can be applied to any receiver

which is capable of bringing the heat transfer fluid *f* to a high temperature under the effect of the concentrated solar radiation.

Three advantageous and particularly adapted embodiments will be cited.

In the first, in keeping with FIG. 3*a* for example, and in the second, the solar receiver **10** comprises a cavity **11**, the cavity **11** being provided with an opening **12** transparent to concentrated solar radiation, and at least one solar absorption element **13** irradiated by concentrated solar radiation through the opening **12**. Opening means any <<window>> which lets through solar radiation, whether glassed or a simple hole in the wall of the receiver **10**. The opening **12** can be enclosed by a collector cone **16** (secondary concentrator) made of reflecting material, as in FIG. 2, this cone concentrating even more precisely concentrated solar radiation to the opening **12**.

The solar absorption elements **13** are refractory elements which will heat under the effect of the concentrated solar radiation. It is at the level of their wall where the solar energy is effectively transformed into heat. The first and the second mentioned embodiment of the receiver **10** differ from the form of these elements **13** and the circulation drawing of the heat transfer fluid *f*.

In this first embodiment, as is evident in FIG. 3*b* and respectively in FIG. 3*c*, the solar absorption element or the solar absorption elements **13** are advantageously ducts and/or tubes lining at least one wall of the cavity **11**, in particular the wall opposite the opening **12**, that is, the wall exposed to the concentrated radiation. In this way, the heat transfer fluid *f* circulates in the solar absorption elements **13**, and it is brought to a high temperature by heat transfer in contact with the internal surface of the solar absorption elements **13**. Alternatively, the ducts and/or tubes cannot be placed against a wall of the cavity **11**, so as to indirectly receive radiation by reflection on the walls of the cavity **11**.

The heat appears on the external wall of these tubes or ducts, and is transmitted by conduction to the internal wall, cooled by passage of the heat transfer fluid. In the <<ducts>> configuration, which are made in the material comprising the base of the cavity **11**, it is noted that there is an external wall common to all the ducts, which is actually the base of the cavity **11**. In the case of parallelepipedic or semi-cylindrical solar receivers **10**, most often the case, this wall is substantially orthogonal to the axis of incidence of the solar radiation, and therefore undergoes the maximal rise in temperature. In the <<tubes>> configuration (in this configuration the absorption elements **13** are formed by tubes separate from the cavity **11**), multiple reflections on the base of the cavity **11** ensure that the entire periphery of the external wall of the tubes is exposed.

The internal exchange surface between the tubes/ducts and the heat transfer fluid per se has an interest in being the highest possible since the coefficients of conducto-convective exchanges with a solid-gas interface are low. It is also not useful to have large diameters of tubes/ducts, since exchanges are made on the wall only. Advantageously, tubes/ducts of small diameter are therefore multiplied (below is the description of a particularly preferred embodiment).

The selected materials are materials ultra-resistant to high temperatures (capable of supporting 2000° C. for several hours), but relatively conductive of heat. The solar absorption elements **13** and more particularly the cavity **11** are therefore selected in ceramic or graphite (sublimation point

at 3652° C.). Tubes made of metallic materials with high-temperature fusion are also feasible, though graphite is preferred.

Since the heat transfer fluid *f* is separated tightly from the cavity **11** by the wall of the tubes, the cavity **11** most often no longer needs to be hermetic. A simple hole is adequate as opening **12**, but advantageously it is still fitted with a glass pane to insulate the cavity **11** from the oxidising atmosphere, harmful in particular to graphite. The cavity can then be filled with neutral atmosphere (nitrogen or argon, for example).

Alternatively, the heat transfer fluid *f* can be circulated directly in the cavity **11**. The opening **12** is obligatorily closed by a porthole **14** transparent to concentrated solar radiation, sealed off and resistant to pressure. In this second embodiment, the fluid *f* circulates around the thermal absorption elements **13**, and not inside the latter. The fluid is therefore brought to a high temperature by heat transfer in contact with the external surface of the solar absorption elements **13**, still irradiated by the concentrated solar radiation via the opening **12**. This solar receiver **10** is evident in FIG. 4.

There is therefore greater liberty of form for the absorption elements **13**, and advantageously beehive structures, porous foam or aerosols are selected which offer a very high exchange surface with fluid *f*. These structures typically occupy the entire cross-section of the cavity so as to force the heat transfer fluid *f* to pass through it.

The preferred materials and the dimensions are substantially the same as for the first embodiment.

According to the advantageous third embodiment, the cavity **11** and the opening **12** can be omitted. The solar receiver **10** is in fact traversed by a duct in which the heat transfer fluid *f* is in motion, the heat transfer fluid *f* being brought to a high temperature directly by heat transfer in contact with the internal surface of a wall of the duct the external surface of which is irradiated by the concentrated solar radiation. To boost heat transfer, elements generating turbulences in this duct can be used. This solution is very close to the first with ducts integrated in the cavity, it must be known that at higher temperatures a cavity will be necessary to decrease heat losses by infrared radiation to the exterior.

Such a solar surface duct receiver is described especially in French patent application FR0957204. Particularly Preferred Embodiment of the Receiver

A particularly advantageous solar receiver **10** developed by PROMES laboratories (Unit CNRS 8521), according to the first embodiment of the receiver described previously, is illustrated in FIG. 5. This multi-tubular experimental receiver was able to heat the heat transfer fluid *f* to a temperature of 2073° K.

This receiver **10** comprises an aluminium envelope and a substantially cubic graphite receptive cavity **11** (around 40 cm per side). The opening **12** letting the concentrated solar flow through has a diameter of 13 cm. The tubes **13**, seven in total, are 800 mm long with an inner diameter of 18 mm and an outer diameter of 26 mm. They are positioned staggered horizontally. A hemispherical porthole **14** made of quartz 360 mm in diameter and 5 mm thick insulates the cavity **11** from the oxidising atmosphere. The zone **15** (overall parallelepipedic, with a side of the order of 800 mm) enclosing the cavity **11** is filled with layers of insulating materials (for example fibrous materials made of aluminosilicate or graphite felt) which help maintain heat in the cavity **11**.

Heat Transfer Fluid and Injection

Fluid means liquid or gas entering the solar receiver **10**, provided however that at levels of high temperature achieved at the outlet of the solar receiver **10** the vaporisation temperature has been attained such that the heat transfer fluid *f* may take the form of a gas jet *g*. Gaseous fluid in the STP (Standard conditions for temperature and pressure) is however preferred as there is no problem of change of state (consumption of vaporisation enthalpy).

Very many fluids can be used, the choice depends essentially on the industrial process for which the fluid reproduces a combustion flame. In fact this fluid must either be stable at nominal high injection temperatures in the chamber **30** and relative to components used during the process, subject to interfering with this process, or on the contrary be an active constituent of the process whereof impact is preferred, for example a reagent of a chemical reaction. Also, fluid must advantageously have good thermal conductivity to rapidly store heat energy and/or have high calorific capacity C_p to store much energy.

In general, dihydrogen (in the absence of dioxygen) and helium are particularly advantageous for their thermal performances and their stability. There are also argon, dinitrogen, and the use of ambient air is also feasible for its low cost and its availability.

For the heat transfer fluid *f* to be injected into the high-temperature chamber **30**, the injection means **20** can comprise pump systems, but only extremely expensive and refined systems can resist temperatures which prevail at the outlet of the solar receiver **10**.

This is why the problem is advantageously reversed by effecting the rise in pressure prior to entry to the solar receiver **10**, when temperature levels are still low: the heat transfer fluid *f* is injected under pressure into the solar receiver **10**. The pressure level must be adjusted to the mechanical resistance of the elements of the receiver, in particular the tubes **13** or the porthole **14**, if there is one. It can nevertheless advantageously be raised by several bars.

The injection means **20** of the heat transfer fluid *f* in the high-temperature chamber **30** consist advantageously of a single conduit via which the heat high-temperature transfer fluid *f* escapes from the solar receiver **10** to the high-temperature chamber **30** under the effect of the internal pressure in the solar receiver **10**. The diameter of this conduit can be adjusted as a function of pressure and fluid flow in the solar receiver so as to adjust the speed of high-temperature gas exiting from the conduit (in other terms the size of simulated flame) optimal for the process. So if Q is the flow, v the discharge speed and S the surface of a section of the conduit, $Q=vS$. Assuming that fluid obeys the law of perfect gases $PV=nRT$, this gives $PQ=DRT/M$, with D the preferred mass flow, and M the molar mass of the fluid. This gives $S=DRT/PvM$. Experimental examples of significant values will be given later, but in general radii of a few centimeters are significant, in particular between 1 cm and 10 cm.

The connection between the solar receiver **10** and the injection means **20** can be done advantageously by means of a collector **21** such as illustrated in FIG. 6, in particular if the solar receiver **10** is of tubes/ducts **13** type. It is evident that a collector **21** is also present on the receiver of FIG. 5.

The collector **21** enables a mixing zone to homogenise the temperature (the configuration of the cavity **11** can have the tubes **13** not all receive the same quantity of energy) before introduction of high-temperature fluid to the high-temperature chamber **30** via the injection means **20** (in particular when the latter comprise a conduit). Graphite is material

adapted by its easy forming capacity and its resistance to temperature. FIG. 6 shows a pointed shape: the hydrodynamic must particularly preferably favour high discharge speeds (between 0.3 and 0.8 Mach are preferred in some industrial processes) to ensure turbulence and efficacious transport of particles. It is understood that the collector **21** must be located closer to the solar receiver **10** to minimise heat losses.

Also, the pressure in the high-temperature chamber **30** is advantageously less than the pressure in the solar receiver **10**. So the form of gas jet *g* is caused by detente of the heat transfer fluid *f* as it leaves the injection means **20**. This detente of Joule-Thomson type is advantageously permitted with the conduit terminating in a simple nozzle. The injection means **20** have an effect comparable to that of a nozzle. They augment the regularity of the gas jet and ease diffusion of its heat energy in the high-temperature chamber **30** for the needs of the process.

The zone **31** of the chamber **30** where the gas is injected advantageously has a relatively central position to best avoid heating the walls. In FIG. 2, the zone **31** corresponds to a premixing zone (see below).

According to the needs of the industrial process, it is quite possible to have several high-temperature gas arrivals for several injection zones **31**, and therefore several fluid injection means **20**. The conduit can branch out.

It is clear that it is possible to select fluid which despite high temperatures prevailing in the receiver **10** is always in the liquid state entering the conduit, and which vaporises instantaneously entering the chamber **30** following detente.

As mentioned previously, the temperature of the heat transfer fluid *f* at high temperature is advantageously between 1000° C. and 2500° C. according to the flame temperature required by the industrial process.

High-Temperature Chamber

As explained previously, the system according to the invention can be adapted to a large number of industrial processes. This adaptation goes through the choice of a high-temperature chamber **30** optionally specific to the preferred process. It is evident that the invention is not limited to any industrial process in particular.

For example, in the fields of metallurgy, steel works or ceramics, a furnace can be selected as high-temperature chamber **30**, the high-temperature industrial process being a process for obtaining metallic or ceramic material, such as decarburising of the iron mineral for steel production (chamber **30** is a blast furnace), or the fusion of silica for the production of glass.

Alternatively, the high-temperature chamber **30** can be a chemical reactor, and the high-temperature industrial process an endothermal chemical reaction.

Chemical Reactor

Use of the solar system **1** according to the invention for performing endothermal chemical reactions is particularly significant due to the possibility of choosing the heat transfer fluid, that is the gas or gases comprising the flame, as opposed to the case of combustions, where the same gases (CO_2 , CO , NO_x , SO_x . . .) are always being produced. The latter are pollutants which inter alia are mixed with the reaction products and contaminate them. So, the heat transfer fluid *f* advantageously comprises chemically inert gas and/or a reagent of said endothermal chemical reaction and/or a product of said endothermal chemical reaction. It is therefore possible to have a perfectly controlled reaction.

At least one reagent *r* of said endothermal chemical reaction can also be injected into the high-temperature reactor **30** at the level of the injection zone **31** of the heat

transfer fluid *f* high-temperature mentioned previously. So, in the case of reaction with two reagents, one can be used as heat transfer fluid, and the second can be injected into the zone **31**. This produces a good mixture whereof one of the reagents is already at a high level of energy. It is also possible to modulate the position of the injection of the reagent (with injections more or less close to the gas flame). This is used in industrial processes to act on the size distribution of particles produced which are submitted to different dwell times.

It is also feasible for there to be more than one high-temperature chamber **30a**, **30b**, etc. each being a chemical reactor, in particular in the case of complex reactions. These reactors **30** can advantageously be installed in series, the products of the *n*th reactor being injected into the *n*+1st reactor.

For example, there is the instance of a reaction $A+B+C \rightarrow D$, which would contain the following sub-reactions (*X1* and *X2* are reactional intermediaries):



There would be three reactors **30a**, **30b**, **30c** each respectively conducting these sub-reactions.

An advantageous embodiment would be using *B* as heat transfer fluid, providing an injection of *A* into the first reactor **30a** and the third reactor **30c**, and an injection of *C* into the second reactor **30b**.

Cracking of Methane

The solar system **1** is particularly adapted to the industrial processes of the production of hydrogen or nanoparticles of carbon from gaseous precursors, often called <<furnace processes>>, or <<procédés au four>> in French.

Accordingly, the preferred endothermal chemical reaction is the cracking of methane for co-synthesis of dihydrogen and carbon black: $CH_4(g) \rightarrow 2H_2(g) + C(s)$, $\Delta H^\circ = 75 \text{ kJ/mol}$

Advantageously, the heat transfer fluid *f* is dihydrogen, methane being injected into the high-temperature reactor **30**. Any other inert gas stable at very high temperature, such as Ar, He, N_2 , can also be used as heat transfer fluid *f* then separated for recycling on leaving the reactor **30**.

If the system according to the invention is particularly adapted, it is because performing the cracking reaction at the level of walls irremediably results in the growth of pyrolytic carbon deposits.

FIG. 7 illustrates a fluid circulation plan (Prosim® software) illustrating particularly advantageously integration of the solar system **1** according to the invention in an industrial installation for production of hydrogen and carbon blacks.

As it leaves the solar receiver **10**, hydrogen is injected in the form of gas jet *g* at 1873°K . This hot gas is then mixed with methane in a high-temperature adiabatic reactor **30**, shown in FIG. 7 by two virtually consecutive high-temperature reactors **30a**, **30b**: the cracking reaction in fact comprises two reactional steps (which take place physically in the same reactor **30**).

In the first reactor **30a**, the so-called coupling reaction of methane is carried out. Two methane molecules <<combine>> into one molecule of acetylene: $CH_4 \rightarrow \frac{1}{2} C_2H_2 + \frac{3}{2} H_2$ (reaction rate 0.9). In the second reactor **30b**, the acetylene is dissociated into hydrogen and carbon $C_2H_2 \rightarrow H_2 + 2C$ (reaction rate 0.9).

The process is restricted such that the output temperature of the reactional zone, measured via a sensor, is not less than

900°C . (minimal temperature of the carbon black production process called <<Thermal black>>) by controlling the recirculation flow of H_2 heat transfer via the electronically controlled 3-way valve **7**. The flow of methane, injected at the level of the inlet **8**, is as such here fixed at 400 kg/h.

The products must then pass through a separator **4a** of cyclonic exchanger type which separates carbon blacks from gaseous products. Before this cyclone **4** is reached, an exchanger **5** (<<Multi Fluid Heat Exchanger>>) advantageously recovers heat from products both for preheating of the reagent and also gas for heating.

After the cyclone **4a**, bag filters **4b** can prove necessary for secondary filtration prior to purifying of gaseous products in a pressure-modulated adsorption column **6** (purity of H_2 100%, recovery rate 95%). Earlier, the gas must be compressed. Some of the hydrogen purified at the level of the column **6** is recirculated as heat transfer fluid *f* to the solar receiver **10** after preheating at the level of the exchanger **5**, the other part is recovered at the level of the outlet **9c** for evaluation (production of 100 kg/h). The power required for solar heating of hydrogen at the level of the solar receiver **10** is 2.5 MW. Carbon blacks recovered at the level of the cyclone **4a** (output of carbon **9a**) and at the level of the filters **4b** (output of carbon residue **9b**) can then be conditioned (production of 300 kg/h).

It should be noted that the dissociation process proposed here is quite similar to conventional industrial processes, where only the heat contribution is different: instead of using the combustion of fossil sources for heating methane, high-temperature fluid previously heated by concentrated solar energy is injected. This leaves the same flexibility as that which the traditional furnace process can exhibit in terms of pre- and post-treatment of products (ex: oxidising post-treatment for adjusting the properties of carbon black).

The system according to the invention is not at all limited to cracking of methane, and the expert can adapt it to execution of any industrial process needing a combustion flame.

Process

According to a second aspect, the invention relates to a process associated with the solar system according to the first aspect of the invention.

This is therefore a process for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises steps of:

- irradiation of a solar receiver **10** in which heat transfer fluid *f* circulates by concentrated solar radiation to bring the heat transfer fluid *f* to high temperature;
- injection of the heat transfer fluid *f* into a high-temperature chamber **30** in the form of a gas jet *g* reproducing a combustion flame;
- performing said high-temperature industrial process in the high-temperature chamber **30** under the effect of the combustion flame produced.

This process reprises the mechanisms explained previously. It applies to the same industrial processes and it is advantageously implemented by one of the solar system embodiments described previously.

The invention claimed is:

- 1.** A solar system for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises:
 - a solar receiver exposed to concentrated solar radiation, in which a liquid or gaseous heat transfer fluid is brought to high temperature;

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at least one high-temperature chamber distinct from the solar receiver in which said high-temperature industrial process is performed;

injection means separating the solar receiver from the at least one high-temperature chamber, such that the heat transfer fluid is configured to pass from the solar receiver into the injection means and from the injection means into the at least one high-temperature chamber in the form of a gas jet reproducing a combustion flame.

2. The system as claimed in claim 1, in which the solar receiver comprises a cavity, the cavity being provided with an opening transparent to concentrated solar radiation, and at least one solar absorption element irradiated by the concentrated solar radiation through the opening.

3. The system as claimed in claim 2, in which the solar absorption element or the solar absorption elements are tubes and/or ducts lining at least one wall of the cavity.

4. The system as claimed in claim 3, in which the heat transfer fluid circulates in the solar absorption element or the solar absorption elements, the heat transfer fluid being brought to high temperature by heat transfer in contact with the internal surface of the solar absorption elements.

5. The system as claimed in claim 2, in which the heat transfer fluid circulates in the cavity, the opening being covered with a porthole transparent to concentrated solar radiation and sealed, and the heat transfer fluid being brought to high temperature by heat transfer in contact with the external surface of the solar absorption element or solar absorption elements.

6. The system as claimed in claim 2, in which the cavity and the solar absorption element or the solar absorption elements are made of ceramic or graphite.

7. The system as claimed in claim 1, in which the solar receiver is traversed by a duct in which the heat transfer fluid is in motion, the heat transfer fluid being brought to high temperature by heat transfer in contact with the internal surface of a wall of the duct whereof the external surface is irradiated by the concentrated solar radiation.

8. The system as claimed in claim 1, in which the heat transfer fluid is injected under pressure into the solar receiver.

9. The system as claimed in claim 8, in which the injection means of the heat transfer fluid in the high-temperature chamber consist of a conduit via which the high-temperature heat transfer fluid escapes from the solar receiver to the high-temperature chamber under the effect of the internal pressure in the solar receiver.

10. The system as claimed in claim 9, in which the pressure in the high-temperature chamber is less than the pressure in the solar receiver, the form of gas jet being caused by the detente of the heat transfer fluid as it leaves the injection means.

11. The system as claimed in claim 1, in which the temperature of the high-temperature heat transfer fluid is between 1000° C. and 2500° C.

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12. The system as claimed in claim 1, in which the high-temperature chamber is a furnace, and the high-temperature industrial process is a process for obtaining metallic or ceramic material.

13. The system as claimed in claim 1, in which the high-temperature chamber is a chemical reactor, and the high-temperature industrial process is an endothermic chemical reaction.

14. The system as claimed in the claim 13, in which the heat transfer fluid comprises chemically inert gas and/or a reagent of said endothermic chemical reaction and/or a product of said endothermic chemical reaction.

15. The system as claimed in claim 13, in which at least one reagent of said endothermic chemical reaction is injected into the high-temperature reactor at the level of an injection zone of the high-temperature heat transfer fluid.

16. The system as claimed in claim 13, in which the endothermic chemical reaction is the cracking of methane.

17. The system as claimed in claim 16, in which the heat transfer fluid is dihydrogen, methane being injected into the high-temperature reactor.

18. The system as claimed in claim 13, comprising a plurality of high-temperature chambers each being a chemical reactor, the products of the nth reactor being injected into the n+1st reactor.

19. A process for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises steps of: irradiation of a solar receiver in which a liquid or gaseous heat transfer fluid circulates by concentrated solar radiation to bring the heat transfer fluid to high temperature;

injection of the heat transfer fluid from the solar receiver into a conduit and from the conduit into a high-temperature chamber distinct from the solar receiver in the form of a gas jet reproducing a combustion flame; and

performing said high-temperature industrial process in the high-temperature chamber under the effect of the combustion flame produced.

20. A solar system for providing volumetric energy reproducing the effect of a combustion flame for a high-temperature industrial process, characterised in that it comprises:

a solar receiver exposed to concentrated solar radiation, in which a liquid or gaseous heat transfer fluid is brought to high temperature;

at least one high-temperature chamber distinct from the solar receiver in which said high-temperature industrial process is performed;

a conduit separating the solar receiver from the at least one high-temperature chamber, such that the heat transfer fluid is configured to pass from the solar receiver into the conduit and from the conduit into the at least one high-temperature chamber in the form of a gas jet reproducing a combustion flame.

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